

Editorial

Hydrogen Peroxide in Adaptation

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Received 31 October 2012; Accepted 31 October 2012

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The perception of the roles of H₂O₂ in living systems has come a long way, transcending from H₂O₂ being considered as (i) exclusively damaging; (ii) a necessary evil “unwanted but an inevitable product of aerobic metabolism”; (iii) important for specific biological processes that involve ROS aggressiveness, such as the battle of the innate immune system with pathogens; and (iv) signalling species [1–6]. The list does not end here. A number of studies have illustrated that at concentrations in the high physiological range, H₂O₂ induces more permanent, modifying changes, adaptations, increasing the resistance of biological systems to the same stimulus (hormesis) or other stressors (cross-adaptation), or enabling the adaptation to altered ecology. The capability of H₂O₂ to induce the synthesis of a large number of proteins and to provide cross-resistance implies that living systems may intentionally produce H₂O₂ as a component of adaptation in response to different fluctuations and perturbations shifting the system away from homeostasis [7–9]. Some data even implicate an important role of H₂O₂ in interspecies communication and in the development of multicellularity [10].

Examples of signalling roles of H₂O₂ emerge at progressing pace, although the field of redox research is yet to take a fully organised and systematised profile. To date it has been shown that H₂O₂ participates in signalling pathways responsible for the regulation of cell differentiation, proliferation, migration, survival and apoptosis, and mitochondrial relocation and of various biological processes, such as vascularisation, angiogenesis, vascular tone control, maintenance of glucose level, oxygen tension regulation, calcium metabolism regulation, immune system control, and wound healing [3–6]. Redox signalling was first described

in *prokaryotes*. An excellent example of an H₂O₂ sensor is the transcription factor OxyR which is activated by H₂O₂ oxidation of one specific cysteine (thiol) residue. The thiol residue is modified to –SOH, –S–SG, or –S–S– (with another cysteine that emerges onto the protein surface due to conformation changes). Activated OxyR targets multiple genes involved in peroxide removal and the regulation of iron metabolism (the latter is aimed at preventing the Fenton reaction). An example of H₂O₂ signalling in bacteria which is involved in the regulation of processes other than redox control is the *Staphylococcus aureus* virulence factor MgrA, the oxidation of which results in increased resistance to antibiotics. A system composed of the enzyme Orp1 and the transcription factor Yap1 is yeast’s analog of OxyR. A specific –SH residue in Orp1 is oxidised by H₂O₂ to –SOH, further reacting with a cysteine on Yap1 to form a disulfide bridge. The consequent change in conformation leads to nuclear accumulation of Yap1 which promotes the expression of a number of redox enzymes. A similar system is found in another fungus, *Schizosaccharomyces pombe*, but with an additional feature that Pap1 (the Yap1 homologue) is not activated if H₂O₂ is present at concentrations exceeding 1 mM. In this case, the critical cysteine residue in the activator Tpx1 (the Orp1 analogue) is oxidised to –SO₂H resulting in Tpx1’s inability to oxidise Pap1 and to provoke the formation of a specific intramolecular disulphide bond in Pap1 [1–3]. Cells and tissues in mammals and other complex organisms are much more relaxed with respect to an external prooxidative environment as they have specialised organs to deal with the hazards. As a consequence, they generally do not need rapid redox responses and are in position to develop redox signalling pathways that are involved in the regulation

of processes other than feedback activation of antioxidative defence.

The basic principles of H_2O_2 signalling in mammals have been described [3–6]. They are as follows. (i) The inactivation of phosphatases. The oxidation of a specific thiol residue to $-SOH$ prevents it from accepting phosphate. In some cases a disulphide bond is formed thus preventing further oxidation to $-SO_2H$ or $-SO_3H$ and irreversible inactivation. (ii) The activation or inhibition of kinases. ASK1 is activated by H_2O_2 -mediated oxidation of two cysteine residues to $-SOH$ which leads to intermolecular disulphide bond formation and multimerisation. In vascular smooth muscle cells, Src kinase is activated by H_2O_2 , which results in cofilin and myosin light-chain kinase inactivation and consequent F-actin stabilisation and promotion of actin-myosin interactions, thus enabling migration. (iii) The activation/derepression of transcription factors. NF- κ B, p53, HIF-1 α , and AP-1 have redox sensitive thiol groups in DNA-binding domains. In addition, it has been proposed that H_2O_2 -provoked oxidation of specific thiol residues in DJ-1 and the Nrf2-inhibitor KEAP1 results in derepression of Nrf2-regulated transcription of a set of enzymes involved in antioxidant and cytoprotective response. NF- κ B is activated by thiol oxidation, while I κ B kinase can be inhibited by H_2O_2 , which together makes a biphasic redox-sensing mechanism. (iv) The activation of ion channels such as ATP-sensitive K^+ channels or TRPA1. (v) The modification of activity of a number of other proteins. Very recent findings show that H_2O_2 is essential for keeping the immune system under control. The oxidation of T-cell surface thiol switches by H_2O_2 results in suppressed activity and proliferation of T cells. In order to activate T cells, dendritic cells release glutathione which is then cleaved to cysteine to reduce $-SOH$ groups back to $-SH$, thus turning the redox switch-on [11]. It is important to note that H_2O_2 signalling pathways are intertwined with the effects of other signalling species, such as NO and CO.

On the other hand, excessive production of H_2O_2 or some other ROS, and a consequential supraphysiological level of oxidation (i.e., oxidative stress), has been related almost to all human diseases that one may think of [12]. Pertinent to this, the isolation of “natural” or the creation of synthetic antioxidants has become a very lucrative activity, which may explain the “explosion” of antioxidative research in the past four decades. However, promising *in vitro* data have not been translated into success in human clinical trials. If we put aside the fact that many researchers in the field neglect important issues such as (i) which ROS is/are overproduced in the particular condition they are trying to treat? (ii) At which site(s) are ROS produced? (iii) Are ROS important for pathophysiology, or do they merely represent byproducts (in other words, a cause or consequence)? (iv) What are the metabolic properties and targets of the proposed antioxidants (e.g., ascorbate seems to exert some beneficial effects in cancer and sepsis treatment via its prooxidative interplay with iron, but not due to its antioxidative effects)?; the key problem is that the importance of redox signalling has not been appreciated [13]. Such careless, almost concept-free approach is giving antioxidants

a bad name and erases the belief of the importance of redox processes in (patho)physiology in the general (scientific) population, thus compromising the entire redox field. This frustrates the experts and represents the subject of concern for authorities. Barry Halliwell and coworkers have made some extreme examples to point out where this carelessness may end up. They have shown that human urine and feces possess reasonable antioxidative capacities [14]. Although those are “natural products,” there seems to be no interest on the market yet. On the other hand, the European Food Safety Authority recently presented negative scientific opinions on a staggering number of health claims that various compounds and products, some of which are widely accepted “antioxidants,” exert antioxidative/beneficial effects in humans [13]. The explanation for *in vivo* ineffectiveness of many *in vitro* antioxidants hides in the fact that redox signalling is too precious for the cells and tissues to allow exogenous meddling. In addition, the intrinsic enzymatic antioxidative system is more efficient than any supplementary antioxidants (of course, if there is no vitamin deficiency), so cells see the later more as a threat to normal redox signalling and less as a help [13, 14]. In order to maintain a flexible redox poise living systems have developed refractory mechanisms [15]. A good illustration of the activity of refractory mechanisms represents the fact that one may consume huge amounts of ascorbate or vitamin E, but the level of these in the blood will not rise above a specific level [15]. When exposed to an excess of exogenous antioxidants, the cells are even prepared to suppress the intrinsic enzymatic antioxidative defence in order to preserve redox homeostasis [16]. From the evolutionary point of view, the refractory system had to be developed or otherwise signalling pathways involved in the regulation of crucial biological processes would have been diet sensitive (e.g., excessive consumption of fruits and vegetables would result in proreductive conditions and the obstruction of redox signalling pathways). Do these facts altogether imply that there is no future for antioxidative therapy? Of course not! We still can adjust the redox milieu with more sophisticated approaches such as the modulation of enzyme and transcription factor activities, or we may offer compounds that have antioxidative effects but are not recognised by cells as antioxidants *per se* (examples being pyruvate, fructose, fructose 1,6-(bis)phosphate, oxaloacetate, fumarate, and metal chelators), to help the organism to fight intracellular oxidative stress against its own “will.”

H_2O_2 is important for the development and differentiation of multicellular communities of unicellular yeast and bacteria and for their adaptation to the ever-changing environment [17]. In yeast, H_2O_2 at moderate concentrations increases longevity. More importantly, H_2O_2 has been proposed to regulate the initial steps of ammonia production which synchronises the development of colonies. Finally, H_2O_2 is involved in the programmed cell death that develops in the centre of a colony. The dead cells in the colony centre are likely to release nutrients that are then used by the younger prosperous cells at the edge of the colony to survive and colonise other localities [17]. It has been shown that many bacterial species express H_2O_2 -producing enzymes L-amino acid oxidase, lysine oxidase, and pyruvate

oxidase [18]. In developing colonies, bacteria seem to be programmed to produce H_2O_2 which cleaves DNA resulting in genetic rearrangements via double-strand DNA break repair and the release of extracellular DNA fragments that are incorporated into the chromosome by competent cells. In this way, bacteria acquire genetic flexibility and advantageous gene variations which enable them to adapt to altered environments during initial phases of the development of biofilms [19]. In multispecies oral biofilms, streptococci use H_2O_2 (at concentrations above 1 mM) during active growth, as a mean of biochemical warfare against competitors, particularly in times of low carbohydrate (food) availability. However, some species recognise streptococcal-produced H_2O_2 as a signal to upregulate the resistance to host innate immune system responses, which also uses H_2O_2 [20]. It is noteworthy that the two host-derived peroxidases in the human oral cavity, salivary peroxidase and myeloperoxidase, use streptococcal-generated H_2O_2 to produce an antimicrobial substance, hypothiocyanite [21]. So in summary what we have here looks like a love/hate redox triangle. It has been documented in complex organisms that exposure to mild oxidative stress initiated by ROS or low level radiation increases resistance to later challenges with higher concentrations or doses of the initial stressor [22]. For example, nematodes exposed to low level oxidative stress cross-adapt, developing increased resistance to other stressors, such as heavy metals [23]. The key players in the adaptive response of nematodes seem to be antioxidative defence, heat-shock proteins and metal binding proteins, which are regulated by a branched gene pattern. It has been reported that nematodes activate intrinsic ROS production when exposed to environmental stress, such glucose deficiency [24]. This implies that oxidative stress/other stressor cross-adaptation may be based upon the role of H_2O_2 signalling in adaptive processes in general. Some insects seem to use H_2O_2 in a rather creative manner in order to survive subzero temperatures. It has been shown recently that the Arctic springtail (*Megaphorura arctica*) suppresses catalase activity and increases H_2O_2 production when exposed to freezing temperatures [25]. Combined with cryoprotective dehydration, this results in the accumulation of H_2O_2 in body fluids to surprisingly high level (15 to 20%, v/v). If we take into account that the mixture of H_2O and H_2O_2 represents a eutectic system, where the freezing point of 20% (v/v) H_2O_2 solution is at -15°C [26], it is clear how arctic insects may use H_2O_2 to prevent deadly freezing in a harsh environment. These findings could also represent the basis for new concepts in cryopreservation research. In mammals, the increased production of H_2O_2 in cardiomyocytes is actively involved in hypoxia-induced preconditioning to ischemia [27]. The preconditioning protects ischemic cardiomyocytes through upregulation of antioxidative defence and by H_2O_2 -provoked opening of mitochondrial K^+ -ATP channels. The importance of H_2O_2 in the development of increased resistance to ischemia by intermittent hypoxia preconditioning is implicated by the fact that treatment with the antioxidant N-acetylcysteine in the preconditioning phase may completely prevent the development of cardioprotection [27]. It has been suggested

that H_2O_2 plays a key role in the endothelial adaptation to exercise by stimulating an upregulation of endothelial NO synthase [28]. Finally, H_2O_2 has been reported to provoke hormesis in mitochondria of pancreatic β cells [29]. An important role in this process seems to be played by the increased expression of mitochondrial uncoupling protein UCP2. Even more, H_2O_2 promotes the secretion of insulin from pancreatic β cells, which generally exhibit low levels of antioxidative defence and are therefore highly redox-sensitive [29]. Complimentary to the cellular level where mitochondria play a dual role (H_2O_2 -producing/energy-converting organelles), there may be a link between the redox and energy status of the living system as a whole.

In this special issue we present you seven (excellent, in our honest opinion) papers providing various examples of adaptive and signalling roles of H_2O_2 . We would like to invite you to find others in the literature, and there will be, for sure, more to come.

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