



# BOOK OF ABSTRACTS

FIRST INTERNATIONAL CONFERENCE  
PROCESSING, CHARACTERISATION AND  
APPLICATION OF NANOSTRUCTURED  
MATERIALS AND NANOTECHNOLOGY  
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First International Conference

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AND APPLICATION OF  
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NANOTECHNOLOGY**

**PROGRAMME  
&  
BOOK of ABSTRACTS**

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NANOSTRUCTURED MATERIALS AND NANOTECHNOLOGY**  
*First International Conference, NanoBelgrade 2012*  
**PROGRAMME & BOOK of ABSTRACTS**

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### **Adsorption and photocatalytic degradation of Reactive Orange 16 dye with hydrothermally modified anatase**

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Anatase is a well-known photocatalyst mostly used to degrade or transform organic and inorganic water pollutants. Its photocatalytic properties depend on surface area, availability of active sites, pore sizes, number and nature of trapped sites, as well as on adsorption/desorption characteristics. In order to reach the highest performance, the main challenge is to find new methods for anatase preparation or modification to achieve a balance between mentioned properties. In this work, hydrothermal treatments of anatase were carried out at 120, 150 and 180 °C during 6, 12 and 18 h in NaOH solution ( $c = 3 \text{ mol dm}^{-3}$ ). The samples were characterized by X-ray powder diffraction (XRPD), scanning and transmission electron microscopy (SEM and TEM). The photocatalytic activity and efficiency of the samples were tested on Reactive Orange 16 (RO16) dye and compared to commercial P-25 powder (Degussa). The powders are coded as UT<sub>t</sub>, where T is temperature of the treatment and t is duration of the treatment. XRPD analysis showed that phase composition of the samples remains the same and confirms absence of the reaction between anatase and NaOH, which would result in formation of a sodium-titanate. Nanoparticles of modified anatase form soft m, as SEM analysis showed  $\mu$ agglomerates that differ in size 1 – 5  $\mu\text{m}$ . Size of modified nanoparticles is about 5 nm what is observed by TEM. It is also found that formation of anatase nanotubes, with lengths 100 – 250 nm, is started. The adsorption power of U120<sub>18</sub> and U150<sub>6</sub> samples is twice as much as adsorption power of P-25 powder. The samples U150<sub>12</sub> and U180<sub>6</sub> have adsorption power very similar to P-25, while all other samples show much lower adsorption power. The photocatalytic activity of all modified samples is lower than photocatalytic activity of P-25, but because of higher adsorption power the overall efficiency of RO16 degradation in water with U120<sub>18</sub> and U150<sub>6</sub> samples is comparable to the efficiency of P-25 powder.