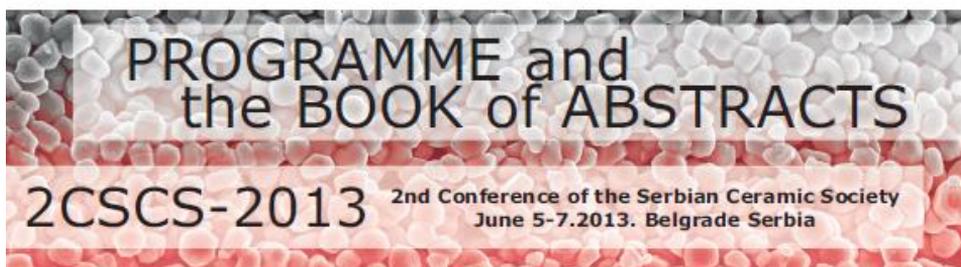


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The Academy of Engineering Sciences of Serbia  
Institute for Multidisciplinary Research - University of Belgrade  
Institute of Physics - University of Belgrade  
Vinča Institute of Nuclear Sciences - University of Belgrade



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Snežana Bošković  
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Zorica Branković

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# **PROGRAMME AND THE BOOK OF ABSTRACTS**

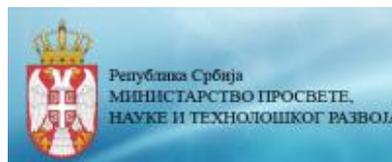
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**Snežana Bošković  
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|---|----|
| <b>C. Stan, C. Andronesu, A. Rusu, J. Pande, C. Munteanu, E.M. Anghel, D. Culita, I. Atkinson, V. Fruth</b><br>SYNTHESIS AND CHARACTERISATION OF Sr-SiO <sub>2</sub> POWDERS WITH ORDERED MESOPORES AND ITS POTENTIAL APPLICATIONS IN DRUG DELIVERY .....       | 62 |
| <b>D. Luković Golić, Z. Branković, N. Daneu, A. Rečnik, G. Branković</b><br>ZnO MESOCRYSTALS FROM SOLVOTHERMAL SYNTHESIS .....  | 63 |
| <b>D. Marinčev, M. Grujić-Brojčin, S. Stojadinović, M. Radović, M. Šćepanović, Z.D. Dohčević-Mitrović</b><br>THE RAMAN SPECTROSCOPY OF TiO <sub>2</sub> /WO <sub>3</sub> COATINGS FORMED BY PLASMA ELECTROLYTIC OXIDATION .....                                 | 63 |
| <b>J. Luković, B. Babić, M. Prekajski, D. Bučevac, Z. Bašcarević, M. Kijevčanin, B. Matović</b><br>SYNTHESIS AND CHARACTERIZATION OF NANOMETRIC TUNGSTEN CARBIDE .....  | 64 |
| <b>J. Ćirković, T. Srećković, K. Vojislavljević, Z. Branković, G. Branković</b><br>STRUCTURAL AND MICROSTRUCTURAL CHARACTERIZATION OF BST CERAMICS OBTAINED BY HYDROTHERMALLY ASSISTED COMPLEX POLYMERIZATION METHOD .....                                      | 65 |
| <b>P.I. Pontón, L. Mancic, S. Savic, B.A. Marinkovic</b><br>FUNCTIONALIZATION OF THE TITANATE NANOTUBES WITH A SILANE COUPLING AGENT .....  | 65 |
| <b>L.M. Nikolić, M. Milanović</b><br>APPLICATION OF THE LAYERED TITANATES IN WATER PURIFICATION ....  | 66 |
| <b>M. Stojmenović, S. Bošković, S. Zec, B. Babić, B. Matović, S. Mentus, M. Žunić</b><br>PROPERTIES OF MULTIDOPED CeO <sub>2</sub> NANOPOWDERS SYNTHESISED BY GNP AND MGNP METHODS .....  | 67 |
| <b>M. Vuković, Z. Branković, D. Poletić, A. Rečnik, G. Branković</b><br>SOLID-STATE SYNTHESIS AND CRYSTAL GROWTH OF PURE VALENTINITE ANTIMONY(III) OXIDE .....  | 67 |
| <b>M. Boskovic, B. Sanz, S. Vranjes-Djuric, V. Spasojevic, G.F. Goya, B. Antic</b><br>ENCAPSULATION OF MAGNETITE NANOPARTICLES IN HUMAN SERUM ALBUMIN .....   | 68 |
| <b>M. Omerašević, U. Jovanović, V. Pavlović, M. Čebela, S. Nenadović</b><br>CESIUM ADSORPTION AND PHASE TRANSFORMATION OF CLINOPTILOLITE  | 69 |
| <b>M. Rosic, J. Zagorac, A. Devečerski, A. Egelja, A. Šaponjić, V. Spasojevic, B. Matovic</b><br>EXAMINATION OF NANOSTRUCTURED Ca <sub>1-x</sub> Gd <sub>x</sub> MnO <sub>3</sub> (x=0.05; 0.1; 0.15; 0.2) OBTAINED BY MODIFIED GLYCINE NITRATE PROCEDURE ..... | 69 |

P-11

**STRUCTURAL AND MICROSTRUCTURAL  
CHARACTERIZATION OF BST CERAMICS OBTAINED BY  
HYDROTHERMALLY ASSISTED COMPLEX  
POLYMERIZATION METHOD**

Jovana Ćirković, Tatjana Srećković, Katarina Vojisavljević,  
Zorica Branković, Goran Branković

*Institute for multidisciplinary research, University of Belgrade, Serbia*

Barium strontium titanate powder  $Ba_{0.8}Sr_{0.2}TiO_3$  (BST) has been obtained by hydrothermal treatment of precursor solution containing titanium citrate, previously prepared by complex polymerization method, and barium and strontium acetates. The powders were calcined at 700°C, pressed into pellets and further sintered at 1280°C using different times (from 1 to 32 h). The phase compositions of sintered samples were followed using X-ray diffractometry and EDS analysis. Microstructural properties were investigated using scanning electron microscopy. It was found that BST sintered samples contained a two-phase structure. Sintered samples underwent an abnormal grain growth, whereby some grains grow faster than the other due to the presence of two-phase structure.

P-12

**FUNCTIONALIZATION OF THE TITANATE NANOTUBES  
WITH A SILANE COUPLING AGENT**

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In order to develop new nanosized filler compatible with the thermoplastic polymer Nylon-11, trititanate nanotubes (TTNTs) were synthesized by standard alkaline hydrothermal treatment of a  $TiO_2$  anatase powder in 10 M NaOH at 120 °C for 24 h. After the synthesis, the as-obtained nanopowder was washed differently (either with water or HCl), in order to prepare TTNTs with high and low sodium contents. Chemical functionalization of TTNTs was performed with 3-aminopropyltriethoxysilane (APTES) coupling agent using two different reaction