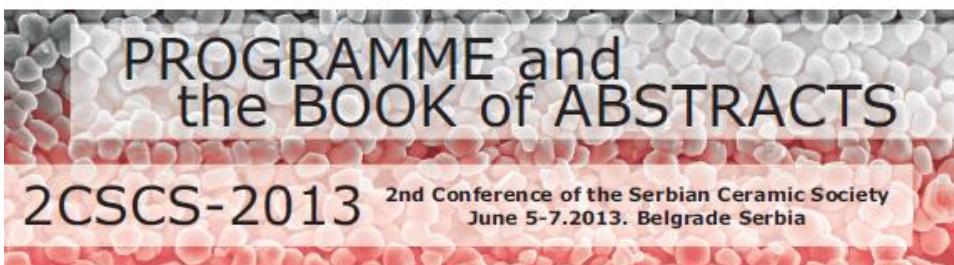


The Serbian Ceramic Society  
The Academy of Engineering Sciences of Serbia  
Institute for Multidisciplinary Research - University of Belgrade  
Institute of Physics - University of Belgrade  
Vinča Institute of Nuclear Sciences - University of Belgrade



Edited by:  
Snežana Bošković  
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Zorica Branković

Programme and Book of Abstracts of The Second Conference of The Serbian Ceramic Society **publishes abstracts from the field of ceramics, which are presented at international Conference.**

***Editors-in-Chief***

Dr Snežana Bošković  
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***Publisher***

Institute for Multidisciplinary Research, University of Belgrade  
Kneza Višeslava 1, 11000 Belgrade, Serbia

***For Publisher***

Prof. Dr Sonja Veljović Jovanović

***Printing layout***

Vladimir V. Srdić

***Press***

FUTURA, Novi Sad, Serbia

CIP – Каталогизација у публикацији  
Народна библиотека Србије, Београд

666.3/.7(048)  
66.017/.018(048)

**SERBIAN Ceramic Society. Conference (2nd ; 2013 ; Beograd)**

Program ; and the Book of Abstracts / 2nd Conference of the Serbian Ceramic Society, 2CSCS-2013, June 5-7, 2013, Belgrade, Serbia ; [organizers] The Serbian Ceramic Society ... [et al.] ; edited by Snežana Bošković, Vladimir Srdić, Zorica Branković. - Belgrade : Institute Multidisciplinary Research, 2013 (Novi Sad : Futura). - 102 str. ; 24 cm.

Tiraž 120. – Registar.

ISBN 978-86-80109-18-3

1. Bošković, Snežana [уредник] 2. Serbian Ceramic Society (Beograd)
- а) Керамика - Апстракти б) Наука о материјалима – Апстракти
- с) Наноматеријали - Апстракти

COBISS.SR-ID 198593292

**The Serbian Ceramic Society**  
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# **PROGRAMME AND THE BOOK OF ABSTRACTS**

**2<sup>nd</sup> Conference of The Serbian Ceramic Society**

**June 5-7, 2013  
Belgrade, Serbia  
2CSCS-2013**

Edited by:  
**Snežana Bošković**  
**Vladimir Srdić**  
**Zorica Branković**

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## **ZnO MESOCRYSTALS FROM SOLVOTHERMAL SYNTHESIS**

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Mesocrystals represent a new class of nanostructured materials, made of crystallographically aligned nanoparticles. Due to their unique structural features they have many physicochemical properties, different from nanoparticulate materials and single crystal materials, which can provide better performance in some applications. Zinc oxide mesocrystals have been synthesized by the solvothermal method at 200 °C during 4 hours from slightly basic (pH = 8) precursor (ethanolic zinc acetate solution in the presence of lithium hydroxide). XRD analysis showed that precursor solution consists of zinc acetate and zinc-hydroxy-acetate. Structural and microstructural properties were analyzed using X-ray diffraction, field emission scanning electron microscopy and transmission electron microscopy. ZnO mesocrystals are hexagonal prisms with diameters of 80 – 200 nm and lengths of 100 – 200 nm, but several larger prisms have a hole in the center. Based on characterization results we have discussed the growth mechanism of ZnO mesocrystals. Dipolar nature of ZnO and planar structure of zinc-hydroxy-acetate with free position of the acetate ions between positively charge planes play crucial role in the formation of the ZnO mesocrystals during the solvothermal reaction.

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## **THE RAMAN SPECTROSCOPY OF TiO<sub>2</sub>/WO<sub>3</sub> COATINGS FORMED BY PLASMA ELECTROLYTIC OXIDATION**

Danijela Marinčev<sup>1</sup>, Mirjana Grujić-Brojčin<sup>1</sup>, Stevan Stojadinović<sup>2</sup>,  
Marko Radović<sup>1</sup>, Maja Šćepanović<sup>1</sup>, Zorana D. Dohčević-Mitrović<sup>1</sup>

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The properties of pure TiO<sub>2</sub> and TiO<sub>2</sub>/WO<sub>3</sub> coatings formed by plasma electrolytic oxidation (PEO) of titanium in 12-tungstosilicic acid water solution have